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Form Approved OMB NO. 0704-0188

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1. REPORT DATE (DD-MM-YYYY)	2. REPORT TYPE		3. DATES COVERED (From - To)		
20-11-2013	Manuscript		-		
4. TITLE AND SUBTITLE			5a. CONTRACT NUMBER		
ORGANIC ESTERS OF {V6O19}; SYNTHONS THAT			W911NF-05-1-0200		
DETECT AND CATALYTICALLY DECONTAMINATE USING AIR		5b. GRANT NUMBER			
			5c. PROGRAM ELEMENT NUMBER 611102		
6. AUTHORS			5d. PROJECT NUMBER		
Daniel A. Hillesheim, Jong Woo Han, Rui Cao, Yurii V. Geletii, Jie Song, Craig L. Hill		5e. TASK NUMBER			
		5f. WORK UNIT NUMBER			
7. PERFORMING ORGANIZATION NAMES AND ADDRESSES			8. PERFORMING ORGANIZATION REPORT		
Emory University			NUMBER		
1599 Clifton Road NE, 4th Floor					
1599-001-1BA Atlanta, GA 303:	22 -4250				
9. SPONSORING/MONITORING AGENCY (ES)			10. SPONSOR/MONITOR'S ACRONYM(S) ARO		
U.S. Army Research Office P.O. Box 12211			11. SPONSOR/MONITOR'S REPORT NUMBER(S)		
Research Triangle Park, NC 27709-2211			47596-CH.29		
12. DISTRIBUTION AVAILIBILITY STAT	EMENT				
Approved for public release: distribution is ur	nlimited				

13. SUPPLEMENTARY NOTES

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14. ABSTRACT

A new class of compounds that detect reducing chemical warfare agents (mustard, HD, and VX) as well as some pollutants and toxic industrial chemicals by reduction of esterified polyoxometalate (POM) units are described. The esterified POM units also catalyze the aerobic (air only) decontamination of these same toxic compounds. The esterified POM units are synthons for supramolecular structures that consequently also detect and catalytically

decontaminate.

15. SUBJECT TERMS

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10. BECCHAIT CERBSHICHTION OI.		- · ·	l .	19a. NAME OF RESPONSIBLE PERSON	
a. REPORT	b. ABSTRACT	c. THIS PAGE	ABSTRACT	OF PAGES	Craig Hill
UU	UU	υυ	UU		19b. TELEPHONE NUMBER 404-727-6611

Report Title

ORGANIC ESTERS OF $\{V6019\}$; SYNTHONS THAT DETECT ANDCATALYTICALLY DECONTAMINATE USING AIR

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ORGANIC ESTERS OF {V₆O₁₉}; SYNTHONS THAT DETECT AND CATALYTICALLY DECONTAMINATE USING AIR.

Jie Song, Daniel A. Hillesheim, Jong Woo Han, Rui Cao, Yurii V. Geletii, and Craig L. Hill*
Department of Chemistry, Emory University, Atlanta, GA 30322

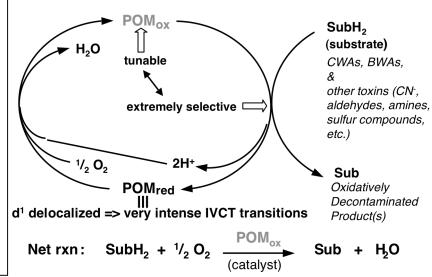
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A new class of compounds that detect reducing chemical warfare agents (mustard, HD, and VX) as well as some pollutants and toxic industrial chemicals by reduction of esterified polyoxometalate (POM) units are described. The esterified POM units also catalyze the aerobic (air only) decontamination of these same toxic compounds. The esterified POM units are synthons for supramolecular structures that consequently also detect and catalytically decontaminate.

INTRODUCTION

Materials that would detect chemical warfare agents (CWAs) and toxic industrial chemicals (TICs) by a simple rapid and dramatic color change are sought. Also materials that can decontaminate CWAs and TICs under benign conditions¹⁻⁴ are attractive. Catalytic decontamination that requires only the ambient environment (air and water vapor at atmospheric temperature and pressure) is highly attractive but clearly very hard to realize.⁵⁻⁷ Two general processes that achieve both detection and aerobic decon are shown in Figure 1. We have recently developed some systems that do both. We report a general synthetic approach to many materials with both capabilities that is based on esterified hexavanadate units (formula $[V_6O_{19}]^{8-}$).

Figure 1 (right). General processes exhibited by POM esters (POM_{ox.red}). Detection involves reduction of lightly colored oxidized POM, top, by several TICs and CWA simulants to form intensely colored, reduced POM, bottom. Catalytic decontamination involves reoxidation POM_{red} to POM_{ox} exposure to air/O₂; net reaction at bottom.



RESULTS AND DISCUSSION

Reaction of tris(hydroxymethyl)aminomethane ("Tris") with decavanadate, (TBA) $_3H_3[V_{10}O_{28}]$, TBA = n-Bu₄N, in dry dimethylacetamide (DMA) under Ar results in production of the bis(trisester) complex, TBA₂[V₆O₁₃{(CH₂O)₃C(NH₂)}₂] in ca. 20% yield. Reaction of functionalized anhydrides, (RCO)₂O, where R = many organic function-bearing groups, with this bis(trisester) POM, TBA₂[V₆O₁₃{(CH₂O)₃C(NH₂)}₂], produces the corresponding derivatized POMs in nearly quantitative yield. One example we address here is the anthracyl functionalized hexavanadate derivative (1) shown in Figure 2.

Figure 2. (right) Structure of complex 1. This readily prepared complex undergoes a dramatic color change when reduced and catalyzes airbased oxidation of the mustard (HD) simulant, 2-CEES.

$$\begin{array}{c|c}
O \\
O \\
O \\
NH
\end{array}$$

$$\begin{array}{c}
V_6O_{19}
\end{array}$$

$$\begin{array}{c}
(TBA)_2
\end{array}$$

Complex 1 on exposure to the mustard (HD) simultant, 2-chloroethyl ethyl sulfide, ClCH₂CH₂SCH₂CH₃ (CEES) turns very dark blue. This color results from the reduced hexavanadate core, and specifically from transitions between states in this POM unit that involve delocalized electrons. This color change is dramatic and rapid even in the presence of small concentrations of CEES and constitutes a simple color-change basis of detection. Simultaneously CEES is oxidized to the corresponding sulfoxide, the HD analogue of which is nearly nontoxic (and far less toxic than mustard sulfone). Similar rapid reductions of the hexavanadate unit are seen with other toxic compounds, that are consequently oxidized.

When the dark blue reduced 1, produced by reaction of CEES, is exposed to the air, the hexavanadate core is reoxidized to the light orange oxidized starting form of the bis(triester) hexavanadate complex. This reoxidation completes a catalytic cycle for decontamination based solely on ambient air (cycle in Figure 1).

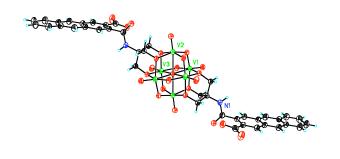
In addition to the color change detection and catalytic air-based decontamination capabilities of 1 and many other accessible structurally analogous bis(triester) V_6 polyanions, these units self assemble into structures that have other attractive properties including porosity. We illustrate this here using just one new example, namely 1. Figure 3 shows the X-ray structure of 1 but also the 3-dimensional porous structure that forms via π -stacking between the anthracene moieties and C-H••(O)POM hydrogen bonding when 1 crystallizes.

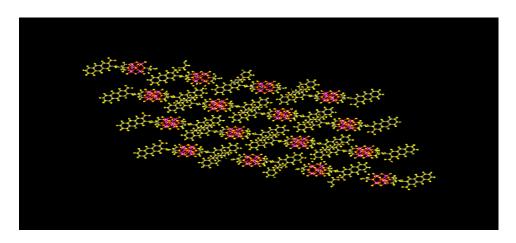
CONCLUSIONS

A new family of functionalized hexavanadate complexes have been prepared that self assemble into a range of porous and other structures that exhibit the properties of the

hexavanadate precursor: color-change detection and catalytic air-based decontamination of reducing CWAs and TICs.

Figure 3. X-ray structure of 1 monomer (right) and the 3D porous structure it forms on crystallization (below).





ACKNOWLEDGMENTS

CLH thanks the U.S. Army Research Office (W911NF-05-1-0200) for funding this work.

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